

PRNC016

PRNC 16

PUERTO RICO NUCLEAR CENTER

NEUTRON DIFFRACTION PROGRAM

Progress Summary Report No. 1

{10 BY UNIVERSITY OF PUERTO RICO UNDER CONTRACT
NO. AT (40-1-183 FOR U. S. ATOMIC ENERGY COMMISSION

---Page Break---

NEUTRON DIFFRACTION

Progress Summary Report No.

March 1963)

Tomsel Alaodévar - Principal Investigator

with Guest Scientist

B. Chalmers Frazer, Helmut J. Bielen,

and M.T. Kay

Puerto Rico Nuclear Center

operated by

University of Puerto Rico

for the

U.S. Atomic Energy Commission

under Contract No. AT-(40-1)-1833,

PRNC 16

---Page Break---

sa.

One of the major uses of a research reactor in basic investi-

gation in the physical sciences is for neutron diffraction studies.

The establishment of a program in this field during the past year is expected to be of key importance in the future development of research in the physical sciences at the Nuclear Center. A brief summary of Program accomplishments in 1964 is presented in this report.

1. Spectroscopy Installation

Since this was the first year of the project, considerable fraction of 1965 was taken up with the construction, installation,

and tests

ting of equipment. One of the Evonucleon-crystal neutron spectrometers to be used in the program went into research operation in January 1963. A view of this machine is shown in Figure 1. A Liquid helium research cryostat is mounted on the couple table. Figure 2 shows a test pattern of powdered iron. This was the first neutron

diffraction data taken in Puerto Rico

Installation of the other spectrometer is expected very soon, This will be a high quality, versatile instrument of the same basic design being used by Brookhaven in its spectrometer construction program for the new high Flux Seam Reactor. The construction is being done at Brookhaven in parallel with that program,

The

electron which is now in use at the Nuclear Center was also required under a cooperative arrangement with Brookhaven. The in-pile collimation, shutter, neutron stop plug, and basic out-pile shielding

were constructed there on a cost reimbursement basis

6. The detector

meter portion was already available and was transferred to the Nuclear Center without charge. This latter portion of the spectrometer is an

old, relatively simple machine, which had become obsolete at

---Page Break---

Figure 1, Neutron spectrometer now in research operation at the

Puerto Rico Nuclear Center. A liquid helium cryostat

is mounted in the sample position.

---Page Break---

-

NEUTRON COUNTS / 50 SECONDS

8

é

8 8 8

r

|

i

{

,

i

\

,

i]

{

28 30

Figure 2.

Me pomderes cytinavical samp.

rau

(200)

re

3234 36 38 40 42 44 48 48 50 52

COUNTER ANGLE °2@

Neutron diffraction cest patcers of powered iron.

f{rst data taken on che specteoneter.

34

the

---Page Break---

wae

Brookhaven, and vas Literally rescued ftom the scrap heap. In

February an excellent sevtron dffractorometer was donated to the

University of Puerto Rico by the Westinghouse Research Laboratories,

4nd on azrangenene {8 being made with the University to replace the older single machine With the nev Yestinghouse unit. The old unte WALL probably be eeanafecced at cost to ABC contract research at the Georgia Institute of Technology

2 caroy

?The structure of Calo, (scheelite) vs

the {rst neutron dif-

fraction search problem undertaken at the tuclear Center. York vas done in collaboration with Dr. ti. I. Kay of the Goorgia Institute of Technology. (1)

?A number of crystals having the tetragonal-scheelite structure hhave been exomined by X-rays. These studies have yielded the heavy ?atom positions, but in most cases the oxygens could be located only from packing considerations, and in all cases the uncertainties in ?the oxygen parameters ere quite large, The discrepancies between the scructures reported for various scheelite-type compounds are

ye than one would expect in an isomorphous series:

Nath neutron diffraction data an accurate direct refinement of all parameters is possible. Starting with oxygen coordinate parameters of Sillen and Åylander,(2) a set of single crystal (KOL)

(1) Paper submitted by H. L. Kay, B.C. Frazer and T. Almodéver for Presentation at the int'l Union of Crystallography, Rome, Italy. Sept. 9-18, 1963.

(2) L. G. Sillen and A. Åylander, Arkiv Kemi, Min. Geol. 17A, no. 4 (1943).

---Page Break---

o3-

neutron data was refined by least squares to yield $x = 0.2617 \pm 0.0007$, $y = 0.1522 \pm 0.0009$, and $z = 0.0861 \pm 0.0002$. These may be compared with the set proposed in the X-ray study of Sillen and Åylander:

$x = 0.25 \pm 0.02$, $y = 0.15 \pm 0.02$, $z = 0.075 \pm 0.015$. Here oxygen is at the 16 (6) general position of space group $I 4_1/a$. An appreciable change is noted in the z parameter, although the neutron value falls within the rather large estimated error limits

3. Balio, and FopSiO,

Preliminary neutron data have been taken on these two compounds using powder diffraction techniques. The magnetic spin structures of these crystals are of principal interest, although the chemical structure of BaFeO_3 is also of considerable interest.

According to the X-ray study of Lander), the Fe^{2+} ion in BaFeO_3

is surrounded by oxygens in approximately square planar coordination.

The oxygen squares share edges to form zig-zag chains in the c direction

of the orthorhombic Caem unit cell. Ordinarily,

one would not expect

Fe^{2+} to have a magnetic moment when in square

planar coordination, but

in this compound the moment appears to have nearly the normal moment

of the divalent ion (approximately $2y$). Lander points out that this is undoubtedly due to the unusually short MI distance in the chains. This distance is only slightly larger than in metallic nickel.

The structure determination of Lander was not carried out with high precision, and the oxygen positions were located mostly on the basis of packing considerations. Neutron data collected at room and liquid nitrogen temperatures did not agree very well with Lander's proposed structure, although preliminary analysis indicates that his

() J. 3. Lander, *Acta Cryst.* 4, 148 (1951),

---Page Break---

-6-

general model is correct. The alternate model one can derive from

packing considerations is definitely incorrect. Adjustment of the

oxygen and barium parameters is now in progress, and preparations

are underway to test

wine the crystal for magnetic order at liquid

helium temperature,

Fep₈i₀, has been studied so far only at room temperature. The observed and calculated intensities for the chemical structure agree fairly well using the parameters of Fe₅Si₁₀, (olivine) are however

some adjustment may be necessary. An accurate crystal structure to

especially important in this case, for only half of the iron atoms are in symmetry-fixed positions, Low temperature experimental studies are planned for the near future.

(4) YL v. Botov, M. Belova, H. Andeianové

Dokl: Akad. Nauk 81,399 (1951).

and P. R. Sufnova;

---Page Break---