

# PRNC142

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PUERTO RICO NUCLEAR CENTER

ACTIVATION ANALYSIS AS A METHOD FOR  
TRACING SUSPENDED SEDIMENTS,

Braulio Mejia Avilés and Knud B. Pedersen

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ACTIVATION ANALYSIS AS A METHOD

FOR TRACING SUSPENDED SEDIMENTS

by

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ABSTRACT

Neutron Activation Analysis was applied to the problem of

judgmentation in the Mayaguez Bay. The method was chosen as an attempt to eliminate the cumbersome chemical methods normally employed, and because it offered the opportunity to increase the sensitivity greatly.

24 analyses were eliminated by

The masking effect of Na

filtering the samples through 0.45 µm millipore filter paper. The

papers were allowed to dry in a closed hood to minimize airborne

contamination and dust from entering the samples. The weight of the particulate matter in the filter paper was determined by weighing the papers before and after filtration.

The filter papers were placed in one inch polyethylene vials and irradiated for 3 sec, together with aluminum standards. They were analyzed 30 sec later, using a NaI (11) crystal. The samples and standards were counted in alternate order for 40 sec live time, for a period which did not exceed 5 min. Aluminum was found to be

present in all the

jeples in different concentrations.

?The results demonstrated that aluminum may be used for tracing  
the sediments contributed by river waters, offering the opportunity of

Jetermining their distribution pattern and settling rate.

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Aluminum Concentrations v. Locations.



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## INTRODUCTION

Many new techniques have developed almost as by-products of nuclear

physics, Activation Analysis is one of them, "Activation" by some type of

nuclear reaction is used to produce a radioisotope from the element to be

analyzed. There are two major advantages in using activation analysis:

first, since the instrumentation available nowadays allows the detection

and measurement of very small amounts of radioactivity, this proves to be

\* A very sensitive method, second, since the radioisotope formed decays

with its own

characteristic radiations and half-life, it is feasible to  
make activation analysis very specific. Besides its high sensitivity,  
activation analysis is a fast and economical method of analysis

Of all the possible nuclear reactions, the neutron reactions have  
been most often used. In the analysis of complex mixtures of nuclides  
the major constituents with high neutron cross-sections.

sections mask the spectra  
of other trace activation products. Most of the methods that are used

to analyze complex mixtures by neutron activation employ suitable chemical  
separations to eliminate the interfering induced activities. If they are  
done before irradiation, some of the interfering activities would be

intensified or new ones would be introduced. In general these chemical

erations are time consuming and therefore, if done after irradiation, are not practical when analyzing for short-lived isotopes.

When analyzing for short-lived isotopes

entirely instrumental

methods must be used. By analyzing the sample immediately after ir-

radiation, very small amounts of these short-lived isotopes can be

measured. In this work the technique was

developed to measure small

amounts of aluminum present in sea water samples.

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[REVIEW OF LITERATURE

Increasing interest has been shown in purely instrumental

methods of neutron activation analysis, and a number of studies not

involving chemical separations have been reported.

(1D jap published a review of some recently developed

v.P. Guinn

Instrumental methods and techniques which are now in use, Some of these methods employ pneumatic tubes to analyze the samples as short as two

seconds after irradiation.

fo. v. Anders(3\*4) as applied the concept of gamma-ray difference

apocrun to suppress the longer-lived components in the gamma spectra of activated samples, The method consists in analysing the samples shortly after irradiation and also after the short-lived isotopes have decayed, by subtracting the second spectrum from the first, the effect of the longer-lived isotopes is almost entirely removed from the gamma-ray spectrum.

?

Dik, Robertson, L.A. Rencitelli and RW, Perkins?) used neutron activation and direct counting techniques to measure concentrations of

numerous trace elements in sea water, marine organisms and pelagic sedi-

ment

Rit, Caldwell) and W.R. Mitchell, Jr. 6 have used rapid neutron activation techniques for quantitative determination of Silicon, Aluminum and Magnesium in rocks. The accuracy and reproducibility of the analyses and interferences from other elements are given,

nit. Greene?? has reviewed the use

of activable stable isotopes

with various types of activation analysis for tracing, Advantages and

disadvantages of the method are given, and several examples of applications

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in medicine, pollution studies and hydrology are described.

P. Marrenoes (20?)

has described the use of radioisotopes in tracing the effluent diversion at a proposed site for sewage disposal, Tracer injections at different sites permit a selection of the most economical site according to the required purification of the sewage.

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When a stable nuclide captures neutrons and forms a radioactive

product which has relatively low absorption cross section, the rate of change of the number of atoms of the radioactive product will be given,

where,  $N_0$  is the initial number of atoms of the radioactive product,

at

where  $\lambda$  is the decay constant of the radioactive product.

where

$N(t)$  = Number of radioactive atoms

$N_A$  = Avogadro's number

$M$  = Weight of the stable nuclide

$\sigma$  = Neutron activation cross section

$A$  = Atomic mass number of stable nuclide

$\phi$  = Neutron flux

$\lambda$  = Decay constant of the radioactive product

multiply both sides of equation 4 by

$\lambda$

$\lambda N$

$\lambda N$

:

$\lambda N$

Then integrating, and assuming  $(0) = 0$

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where  $A_N(t)$  is the activity of the radioactive product. The activity

after irradiation time  $t$  and a decay time is given by



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$\text{pet (eye)} = \text{BREE GQ } \odot$

and the count rate will be,

$\text{Gok. (yt)} \odot \text{det CE) o}$

where  $\odot$  is the

iciency of the detector. For # point source on

the axis of a cylindrical scintillator as in Fig. 1 it is given by,

cs Sone eo ao

where  $G_o$  is the solid angle subtended by the crystal measured from

the point source, and  $\mu_{\text{tot}}$  is the total absorption coefficient.

Quantitative Analysis

In order to perform quantitative analysis of any sample using activation analysis, we can measure the flux, cross section and efficiency of our system, or compare our samples with known samples of standards. In order to compare the activity of the sample with that of a standard, they must have been exposed to the same neutron energy spectrum and neutron flux, and have similar count rates, thus producing essentially the same detector dead time. To obtain the same detector efficiency, the geometries must also be equal.

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ILLUSTRATION FOR CALCULATING EFFICIENCY

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Fig. 2

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by irradiating the sample and a standard for the same time

and essentially in the same place we can assume the same value of the

flux and the same neutron energy spectrum. By choosing the weight of

the standards so that their count rate

will be similar to that of the

samples, similar dead times can be obtained. If the standard and the

sample are located in the same place relative to the detector,

and

having essentially the same dead time the same efficiency can be assumed.

Taking the ratio of the count rate of the sample to that

of the standard we get,

tens = Eta woot ne ey a

stp SMA Ksrmetca - e7Myg AV A

Therefore, if the decay time is the same

where  $W_{ir}$ , is the weight of the standard.

THE EXPERINE!

Reser

-iption of Electronic equipment

1, Scintiitation Detectors

ay

2p

as

4.2" x 2" sodium iodide crystal activated with thallium was

used for the detection of gamma rays

It was coupled to a 10 stage,

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photomultiplier tube which was operated at 935 volts.

## 2, Preamplifier

The purpose of the preamplifier is to increase the signal amplitude from the detector and provide impedance transformation; that is, it transforms the voltage which is developed across the small capacitance at the input of the preamplifier into an approximately equal voltage across the high capacity output cable. Canberra Industries Model 805 Scintillation Preamplifier was used for these experiments,

It is a charge

sensitive, all

Jilicon transistor device which integrates  
cue charge output signals from seintillation/photonultiplier detectors,  
for presentation to the pulse shaping main amplifier,

### 3. Amplifier

?The function of this instrument is to increase further the  
?sinnal amplitude from the preamplifier. The gain can be adjusted to  
obtain the desired energy scale in the analyzer. C.1, Model 810 double  
delay anlplifier was used for this experiment, It accepts the pre~

avplified signals from proportional counters, scintillation

or seniconductor detectors. It yields in turn output signals suitable  
for single channel or rultichannel analysis, and for leading cdre of

ing timing.

#### 4 High Voltage D.C. Supply

This instrument transforms the 115 Volt, 60 Hz A.C.

voltage to high D.C. Voltage to be used by the photomultiplier tube,

the one used for this experiment was @ Fluke Model 409 A, whose output

ranges to 1,500 volts D.C. with Less than .02% ripple and Ina. output.

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#### 5. Input and Power Supply

Ortec Model 401 A Modular System Bin was used to provide

power for the amplifier. It can provide plus or minus 12, or 24

volts D.C. and 115 A.C, delivered to assigned module connector pins.

#### 6. Multichannel Analyzer

The Multichannel Analyzer consists of the following elements:

(1) the analog to digital converter that associates each input signal

with a specific amplitude channel, (2) the memory, of data

which contains the information of the number of pulses in each channel,

and (3) the oscilloscope, typewriter, and curve plotter which provide

For the display of the data which is stored in the memory. The system

for this experiment was a Nuclear Data series 2201 system analyzer, which

is a completely modular multichannel analyzer

A block diagram of the electronic equipment is shown in Fig. 2.

All the irradiations were done at the T.R.N.C. research reactor at a

power level of 1.

Transfer System

1, Description of the system

The system consists of a pipe through which the "rabbit", (the vehicle in which the samples are placed), travels to and from the reactor. The time of irradiation can be adjusted from one second up to thirty minutes, when the rabbit dispatch button is pressed, vacuum is applied at the reactor end of the pipe, thus pulling the rabbit into the reactor. When vacuum is applied instead at the other end



of the pipe the rabbit is made to return.

## 2, Modification Done to the System

Fig. 3 illustrates the modification done to the pneumatic

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MODIFICATIONS DONE TO THE PNEUMATIC system

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Short Pipe

Connection

Detector Shiels

Fig 3

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2

transfer system. When the system was set on "rabbit return" the vacuum applied above the receiving box to make the rabbit return from the reactor also pulled the rabbit back after it had passed the vacuum connection, this made the rabbit lose almost all of its speed making

it impossible to reach the end of the pipe. In order to remove the vacuum effect a butterfly valve was installed in such a manner that it was activated by the rabbit as it approached the vacuum connection. By installing a curved pipe below the receiving box the rabbit was transferred from the reactor directly into the detector shield. A short pipe inserted inside the receiving box allowed the rabbit to bypass

the box.

The detector shield used for this experiment is shown in Fig. 4.

It was made from solid lead in the form of a hollow cylinder. The circular walls are four inches thick and have three circular

cavities, two for the detectors and one for the rabbit tube. The two detector cavities which extend beyond the outer wall of the cylinder provide space for detectors of different sizes. The walls of the external cavities are two inches thick and two lead plugs of the same thickness are located behind each detector.

The inner walls of the shield were covered with .027" of cadmium

and .0625" of copper to absorb the characteristic lead X-rays, which

are produced in the walls of the detector shield,

Since the shield was coupled to the pneumatic system, it had to be sealed air tight, so that the vacuum would effectively pull the rabbit from the reactor.

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Two styrofoam layers were placed inside the shield to provide support for the detectors. The detector was embedded in the top layer thus remaining in a fixed position.

Figure 5 shows the dimensions of the shield and the relative position of the detector.

#### D. Sample Collection and Preparation

Seawater samples were collected using the sample collector

shown in Fig. 6. Glass bottles were previously washed with 4N Nitric

Acid and distilled water, by holding the sample collector at the desired

depth a bottle placed in the sample collector will be completely filled



with Use surrounding water, Samples were collected at different locations

and at different depths in the vicinity of « sewage outfall in the Navaquee Bay.

Since the purpose of this project was to develop @ technique,

so that the samples could be irradiated and analyzed without chemical separations, 1 ml. samples of the water were irradiated, using anal polyethylene vials, and were then directly analyzed. The water samples were irradiated for three seconds and analyzed after thirty seconds of decay. Only Sodium Chloride could be detected as can be seen in Fig. 7 here one such spectrum is shown.

The remainder of the sea water samples were filtered, using 0.45  $\mu$ m millipore filter paper, measuring one inch in diameter. The filter papers were allowed to dry in a closed hood to minimize airborne contamination and dust from entering the samples. The weight of the particulate matter in the filter paper was determined by weighing the

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DIAGRAM OF THE DETECTOR SHILO

DETECTOR

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papers before and after filtration.

Sen

nee of Analysis

?The filter papers were folded and placed in cylindrical

polyethylene vials, one inch long and 1/2 inch in diameter where they

were irradiated for 3 seconds together with vials containing aluminum

standards, The aluminum standards consisted of small pieces of

aluminum foil which were accurately weighed. Both th

samples and

standards were located at the bottom of the vials in order to sustain the geometry relationship with the detector. They were analyzed in alternate order for 40 sec live time, for a period of time which did not exceed five minutes.

The multichannel analyzer had been calibrated to read from 0 to 3 MeV in the four sections of the memory using an aluminum standard.

identified and integrated from 6 channels

The aluminum photopeak was identified from 6 channels before to channels after the peak channel. The result of these integrals were then plotted on semi-Log paper.

Since the decay times have to be equal in order to make

equation 16 valid, the ratio of the integrals have to be taken at a fixed decay time, For this work the value of the integrals 2 nin, after irradiation was taken, hv substitution in equation 16 the weight of aluminum in the samples was determinee.

## Results

The results of this work show that the amount of aluminun present in the particulate matter suspended in sea vater can be measured using instrumental neutron activation analysis. A summary of the results is presented in Table 1.

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A typical representation of the  $A1^{226}$  photopeaks (1.78 rev) obtained from the activation of the particulate matter, as compared to that obtained from the aluminum standard is illustrated in Fig. 8. The

First and third correspond to the suspended sediment sample while the

second and fourth were obtained from the aluminum standard.

In Fig. 9 the integral over the  $A1^{226}$  peak for the sample and the standard is plotted as a function of the decay time.

Similar graphs were obtained from all the samples of particu-

late matter and the amount of aluminum was found to be dependent on location and depth. The area where the samples were collected, and the collection points, are shown in Fig. 10.

In Fig. 11 the concentration of aluminum is plotted for

different locations and depths. The shape of the curve for samples taken at a depth of 1 meter shows that the waste discharged at the sewage outfall and in particular the suspended matter carried by the Yaguez River contain large amounts of aluminum. Curve segments which are also plotted in Fig. 10 show that the aluminum concentration increases with depth.



This effect is more significant in the vicinity of the Yaguez River where

rapid settling of the sediment carried by the river is occurring.

the fact that sulfur and chlorine exist in sea water in high

concentrations constitutes the first foreseeable problem when analyzing

marine samples. when the samples are activated the washing effect of

the sample may not allow the detection of the =

?This problem was successfully overcome by filtering the water samples with

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2

[ DECAY CURVES FOR SAMPLE AND STANDARD

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t 2 s = 3

DECAY TIME (min)

Fig. 9

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APPROXIMATE LOCATION OF SAMPLING STATIONS

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sewage outfall

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OG n wfilipore filter paper. During fileration the particulate watter which fs not in solution is retained in the Filter paper vile the Large ?Srousts of sodium and chlorine which are in solution go through. thie Seisced to @ negligible Level the interference of these two eajor con ?titwents, thus walking possible the detection of cnall amounts of other active products.

?ten weing standards to perform quantitative analysis by

?eutron activation, one of the sost inportant considerations is the fact ?hat the shape and the size of the standard and of the sample should be Sinttar, Tor very snolt samples the variations in efficiency due to the Siference in shape and size of the samples and standards és only aige

nificant when el

Y are located close to the detector. \hen they are laced of a distauce of 10 em or wore the éfference in efficiency

becomes negligible(!2),

Another important consideration in the use of instrumental neutron activation analysis is the choice of irradiation and decay times. These should be chosen so that the ratio of the activity of the irradiated radioisotope, to the activity of the radioisotope which would create the greatest interference, is a maximum. Calculations done to optimize these two parameters for this experiment show that for

measuring plutonium in the presence of sodium the shortest possible

irradiation and decay times

of irradiation and decay should be used

for the radioisotope which is

being produced has a short half-life like for example  $^{24}\text{Na}$

( $t_{1/2} \approx 2-3$  min.) another consideration in the choice of the irradiation

time should be to produce enough  $^{24}\text{Na}$  to be able to measure its activity

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2%

during at Least two half-Lives.

A considerable effort has been devoted to the study of these reactions that take place at the riverocean interface wich modify the couposition of the near shore waters. If the sedinent coming from the rivers can be identitied at the ecean-river interface, the pattern of sedimentation at the river nouth ean be determined, In order to igenti- fy the sedinent which is coning from the river, a tracer must be used,

?the fact that the concentration of alurinus in the suspended solide waa founé to inerease in the vicinity of the Yaguer River, sus

nents the poseibility that aluninun may be used ae a tracer for the

river sediments, Since aluninun is an interral part of the river sedi

pent, ite rate of settling should be the sane as that of the river

one of the advantay

1s of knowing the pattern of distribution

and settling, rate of river sediments is that the blocking effect which

is developed at the mouth of the river could be predicted, and pre-

ventative rei

asures could be developed. It is of fundamental importance

that the river waters are allowed to flow freely into the ocean, since

the limitation of the water flow not only endangers its quality but may

cause flooding during seasons of high precipitation.

?The technique which has been developed here is

igned to use

aluminum as a nat

rally occurring tracer, In situations where aluminum



can not be used as the tracer other tracers can be developed by changing the times of irradiation and decay of the samples.

A few improvements should be made to the water sampling technique,

Since for this research project the greatest effort was devoted

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to the laboratory analysis, the determination of the location where the sample was collected was done only visually. A more precise method

would be used if a study of the ocean

water interface were to be conducted. Also in order not to disturb the sample at the point of collection, a sample collector which could be opened and closed at the desired depth should be used.

conclusion

From the results of this work the following conclusions can be

dram:

by irradiating 1 l of sea water and analyzing it after decay

times up to 30 see only sodium and chlorine can be detected.

2. by filtering the water samples using 0.45  $\mu$  millipore filter

paper and analyzing the filter paper by neutron activation, the sodium

and chlorine interference is reduced and the amount of aluminum present

in the particulate matter suspended in the water can be measured.

3. The concentration of aluminum in the Mayaguez Bay area which

was studied, was as much as one hundred times greater than the value

for standard sea water.

The concentration of aluminum and the weight of particulate

matter per liter of water, increases in the vicinity of the sewage

outfall and near the mouth of the Yaguez River.

5. The concentration of aluminum and the amount of particulate matter suspended in the water also increase with depth.

4. The method and techniques developed in this work can be used to trace the distribution and settling rate of the particulate matter

contributed by certain river waters.

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